

CYCLOADDITIONS OF SUBSTITUTED BENZOPYRAN-4-ONES TO ELECTRON-RICH DIENES: A NEW ROUTE TO XANTHONE DERIVATIVES

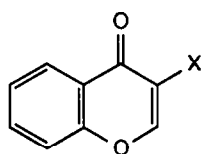
Peter J. Cremins, Suthiweth T. Saengchantara, and Timothy W. Wallace*

Department of Chemistry and Applied Chemistry,
University of Salford, Salford M5 4WT, U.K.

(Received in UK 5 May 1987)

Abstract: The benzopyranones **1** and **3** reacted with 2,3-dimethyl-1,3-butadiene in the presence of titanium (IV) chloride to give the corresponding (4 + 2) cycloadducts **8** and **11**, the former undergoing facile deformylation to give **9** and **10**. Compounds **1**, **3**, and **4** underwent efficient uncatalysed cycloaddition to 1-methoxy-3-(trimethylsilyloxy)-1,3-butadiene **12** to give the respective adducts **13**, **14**, and **18** as mixtures of C-1 stereoisomers. Heating the 3-arylsulphonylchromone **5** with the diene **12** afforded 3-hydroxyxanthone **23** in 50% yield, the presumed cycloaddition - elimination sequence constituting a new route to xanthone systems. Desilylation of **13**, **14**, and **18** in acidic media provided **25**, **26**, and **27** respectively.

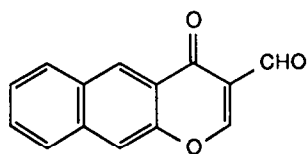
Chromones bearing electron withdrawing substituents at C-3 are highly functional molecules, capable of reacting as Michael acceptors^{1,2} or heterodienes,³ and undergoing various rearrangements on treatment with nucleophiles.⁴ However, their use as 2 π components in (4 + 2) cycloadditions has received surprisingly little attention⁵ despite their obvious potential in such a rôle. We have therefore studied the reactions of 2,3-dimethyl-1,3-butadiene and 1-methoxy-3-(trimethylsilyloxy)-1,3-butadiene (Danishefsky's diene) with the representative pyranone systems **1** - **5**. Our results, herein described, confirm that



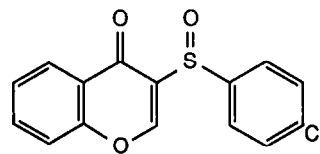
1 X = CHO

2 X = CO₂H

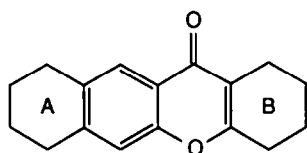
3 X = CO₂Me



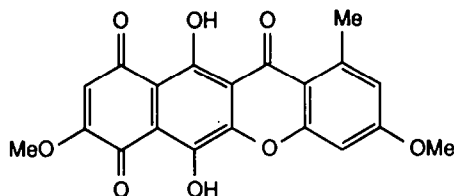
4



5



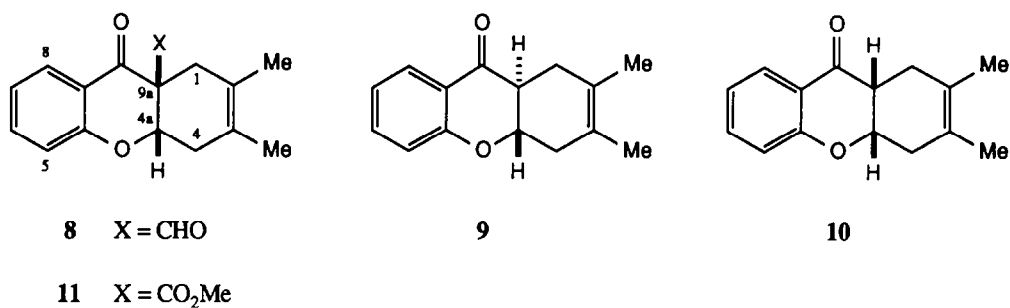
6



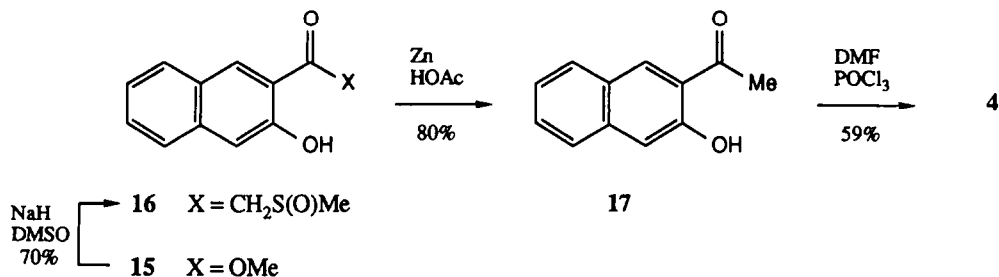
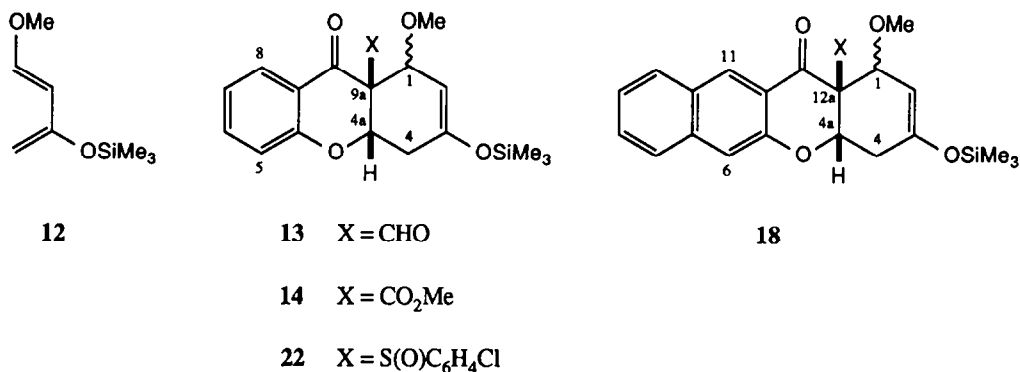
7

such compounds can be used as dienophiles, and that their cycloadditions offer a new synthetic route to heterocycles of the general form **6** (aromatic ring A or B or both). Compounds of this type have potential applications in approaches to xanthone-based analogues of the anthracycline antitumour agents,⁶ or the antiprotozoal and antitumour benzo[*b*]xanthone derivative bikaverin **7**.⁷

Initial experiments were carried out using 2,3-dimethyl-1,3-butadiene. On treatment of the aldehyde **1** with a 10-fold excess of the diene in refluxing dichloromethane, no reaction could be detected after several days, while the inclusion of a catalytic amount of titanium(IV) chloride in the solution resulted in the formation of a complex mixture as judged by thin layer chromatography (t.l.c.). It was suspected that the initial adduct **8**, a tertiary β -ketoaldehyde, would be prone to deformylation under the conditions of the reaction; indeed, increasing the reaction temperature (so as to encourage the deformylation process) produced a cleaner mixture from which the tricycles **9** and **10** (total 41%) were isolated. With the acid **2** similar problems were encountered, the reaction requiring TiCl_4 catalysis and producing a complex mixture of products. However, under the same conditions the ester **3** and 2,3-dimethyl-1,3-butadiene cleanly gave the stable cycloadduct **11** (84%).



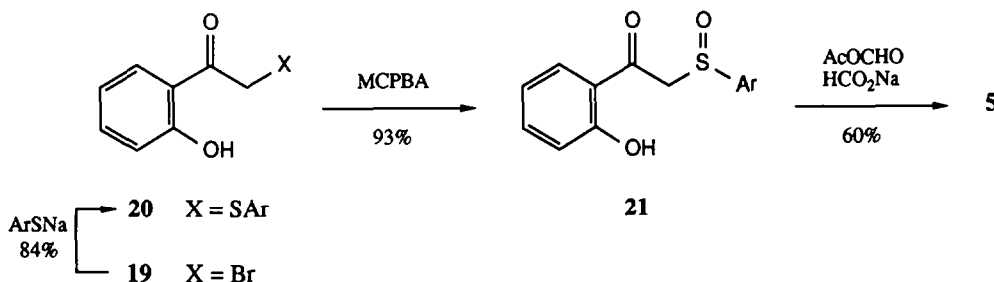
Cycloadditions of **1** and **3** with Danishefsky's diene **12** proceeded at significantly different rates but did not require Lewis acid catalysis. Using a 5-fold excess of the diene at room temperature in dichloromethane, the aldehyde **1** gave an essentially quantitative yield of the isomeric cycloadducts **13** after a few hours, as judged by ¹H n.m.r. spectroscopy. Integration of the signals attributable to the methoxyl (δ 3.1 and 3.3 ppm) and aldehydic (δ 9.9 and 10.0 ppm) protons of each adduct indicated that there had been no useful stereoselection during the cycloaddition (isomer ratio \leq 1.5:1). With the chromone ester **3** the diene **12** furnished a similar mixture of adducts **14**, although in this case the reaction took several days



Scheme 1

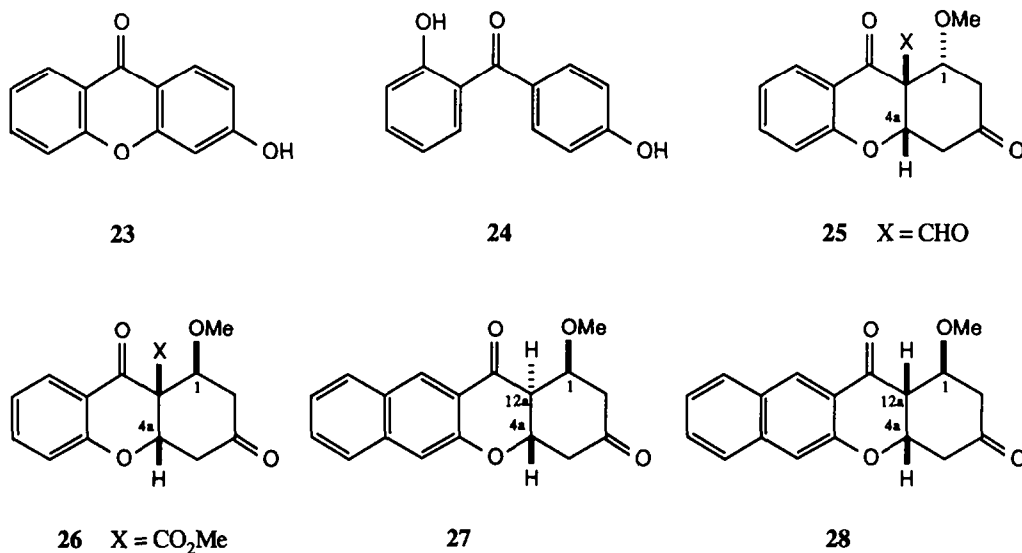
to go to completion. The reaction of the naphthopyran aldehyde 4, prepared using established methods as shown in Scheme 1, with 12 was entirely analogous to that of its homologue 1, producing within a few hours a mixture of the isomeric cycloadducts 18.

It was important to determine whether 3-arysulphonylchromones could function as dienophiles, since their availability in optically active form² makes them potentially useful for chiral induction purposes. The reaction of the sulphoxide 5, prepared as outlined in Scheme 2, with the diene 12 was therefore examined. Using the conditions described above resulted

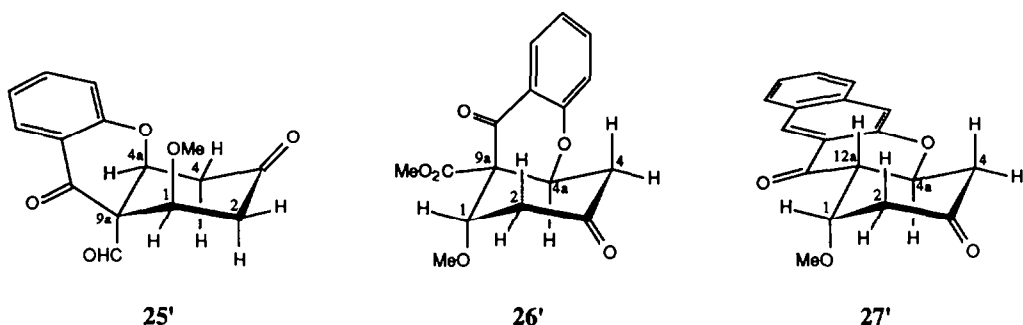


Scheme 2 (Ar = 4-Chlorophenyl)

in the slow (*ca.* 12 days) disappearance of the starting material. The instability of the products precluded their isolation and characterisation, but strong evidence for the formation of the expected adducts 22 was obtained by repeating the reaction at 85 - 90 °C. Under these circumstances the sulphoxides 22 were expected to aromatise *via* the *syn*-elimination of chlorobenzenesulphonic acid and 1,4-elimination of methanol, and indeed after 18 hours the xanthone 23 was isolated from the reaction mixture in 50% yield. This presumed cycloaddition - elimination sequence constitutes a convenient new route to xanthone derivatives.



Although they decompose on standing at room temperature, the cycloadducts 13, 14, and 18 could be desilylated to give stable derivatives. Treatment of 13 with tetrabutylammonium fluoride gave the benzophenone 24 (44%), a result which again reflects the ease with which tertiary β -ketoaldehydes undergo deformylation. In this case the rather basic desilylating agent also provokes elimination and hence aromatisation. Such eliminations were minimised on desilylating with aqueous HF in acetonitrile.⁸ Treatment of the mixture of adducts 13 with this reagent gave several products of which 25 (45%) was isolated; the sensitivity of the latter compound to base was confirmed by heating it with ethanolic sodium acetate, which gave 24 in quantitative yield. The ester adducts 14 gave a mixture on treatment with aqueous HCl in tetrahydrofuran, the product isolated in this case being 26 (22%), while flash chromatography of the complex mixture obtained on treating 18 with the HF



reagent yielded a deformed compound thought to be **27** (45%). The relative stereochemistry of **25** and **26** was evident from their respective ¹H n.m.r. spectra. In that of the aldehyde **25**, the signals due to H-1 (4.51 ppm) and H-4a (5.17 ppm) appear as narrow multiplets, and the fact that neither incorporates a large coupling to its adjacent methylene system indicates that neither is axially disposed. Decoupling reveals that there is also a 'W'-coupling of 1.9 Hz between these methines, confirming their diequatorial arrangement as depicted in **25'**. The spectrum of the ester **26** includes a narrow triplet due to H-1 at 4.59 ppm, but the signal due to H-4a at 5.47 ppm displays a 10 Hz coupling to one of the H-4 methylene hydrogens; these features are consistent with H-1 and H-4a being equatorial and axial respectively, as in **26'**. In the n.m.r. spectrum of **27**, the signal due to H-12a at 2.94 ppm (dd, *J* 2.1 and 12.5 Hz) is consistent with a 1,2-diaxial relationship with only one of the two adjacent methine hydrogens H-1 and H-4a. This indicates a *trans* relationship between these two, but their respective chemical shifts (4.59 and 4.89 ppm) are a little too close for the assignment to be unequivocal. However, reversal of this assignment would require that the structure be **28**, which with the *cis* ring fusion seems unlikely for thermodynamic reasons. The stereochemistry of the product is therefore assumed to be as indicated in **27'**.

From a synthetic viewpoint the results with Danishefsky's diene **12** are particularly significant, since Stoodley and his coworkers have recently developed a diastereoselective method for constructing the saturated A-ring of anthracyclines *via* carbohydrate-derived analogues of this diene.⁹ With an appropriate naphthopyranone dienophile, it may be possible to use a similar protocol in approaches to xanthone-based analogues of these anticancer antibiotics, while the use of a chiral arylsulphinyl activating substituent on the dienophile offers a complementary route to such systems.

EXPERIMENTAL

Melting points are uncorrected. Unless otherwise stated, i.r. spectra were of liquid paraffin mulls on sodium chloride plates, recorded on Perkin-Elmer 297 or Pye-Unicam SP3-100 spectrometers. ¹H N.m.r. spectra were measured for solutions in deuteriochloroform unless otherwise indicated, with tetramethylsilane as the internal standard, on Varian EM 360 (60 MHz), Varian CFT-20 (80 MHz), Perkin-Elmer R32 (90 MHz), or Bruker AC300 (300 MHz) instruments. Mass spectra were measured on a Kratos MS30 instrument with a 70 eV electron impact source unless otherwise indicated, and the peak abundances are quoted as a percentage of the base peak.

2,3-Dimethyl-1,3-butadiene (Aldrich) and 1-methoxy-3-(trimethylsilyloxy)-1,3-butadiene **12** (Lancaster) were used as supplied. Other materials and solvents were routinely purified by conventional techniques.¹⁰ Organic solutions were dried using anhydrous magnesium sulphate and concentrated by rotary evaporation. T.l.c. was carried out on Camlab Polygram SIL G/UV₂₅₄ silica gel plates, and preparative column chromatography¹¹ was carried out using 60H silica gel (Merck 9385). Compositions of solvent mixtures are quoted as ratios of volume. 'Petroleum' refers to a light petroleum fraction, b.p. 40 - 60 °C, unless otherwise stated.

4-Oxo-4H-1-benzopyran-3-carboxylic acid 2. - A stirred mixture of chromone-3-carboxaldehyde **1**¹² (0.5 g, 2.9 mmol), sulphuryl chloride (0.25 ml, 3 mmol), and a few crystals of 2,2'-azobis(2-methylpropionitrile) in carbon tetrachloride (25 ml) was heated under reflux for 3 h. The cooled mixture was evaporated to dryness under reduced pressure and the residue treated with water (30 ml). Filtration, washing with water and ether, and drying under suction gave the *title compound 2* (0.37 g, 68%) as a colourless solid, m.p. 201 - 203 °C (dec.) (ethyl acetate) (lit.¹³ 204 - 205 °C), identical (i.r., n.m.r.) to material prepared by the published method.¹³

Methyl 4-oxo-4H-1-benzopyran-3-carboxylate 3. - A mixture of the acid **2** (1.52 g, 8.0 mmol) and 3% methanolic hydrogen chloride¹⁴ (80 ml) was stirred at room temperature for 24 h. The solution was concentrated under reduced pressure and then diluted with dichloromethane (50 ml). The mixture was washed with water, saturated aqueous sodium hydrogen carbonate, and brine, dried, and filtered. After evaporation the residue was chromatographed [dichloromethane - ethyl acetate (19:1)] to obtain the *title compound 3* (1.43 g, 88%). Crystallisation from ethyl acetate - petroleum ether (1:3) gave a sample with m.p. 93 - 94 °C (Found: C, 64.6, H, 3.85; C₁₁H₈O₄ requires C, 64.7, H, 3.95%); ν_{\max} 1700, 1650, and 1610 cm⁻¹; δ (60 MHz) 4.0 (3 H, s, CO₂Me), 7.25 - 7.5 (3 H, m, 6-H, 7-H, 8-H), 8.3 (1 H, dd, *J* 2.5, 8.5 Hz, 5-H), and 8.74 (1 H, s, 2-H); *m/z* 204 (M⁺, 39%), 173 (90), 146 (100), and 121 (78).

Preparation of 4-Oxo-4H-1-naphtho[2,3-c]pyran-3-carboxaldehyde 4 (Scheme 1).

2-Hydroxy-3-[(methylsulphonyl)acetyl]naphthalene 16. - Using established methodology,¹⁵ a vigorously stirred solution of sodium hydride (100%; 1.4 g, 60 mmol) in dimethylsulphoxide (60 ml) under argon was heated in an oil bath (75 - 80 °C) for 6 h. The resulting solution was cooled in an ice bath and, with vigorous stirring, treated with methyl 3-hydroxy-2-naphthoate **15**¹⁶ (4.04 g, 20 mmol). The resulting brown solution was stirred at room temperature for a further 0.5 h and poured into ether (200 ml), whereupon an oily solid separated out. The ether layer was decanted off and the residue triturated with a small quantity of ethanol and more ether to give a filterable yellow solid. This was collected, dissolved in water (50 ml), and treated dropwise with glacial acetic acid to precipitate out the desired sulphonyl compound **16**, which was collected by filtration and dried *in vacuo* over silica gel. The material thus obtained (3.47 g, 70%) had m.p. 186 - 188 °C, and was used without further purification.

2-Acetyl-3-hydroxynaphthalene 17. - The crude sulphonyl compound **16** (2.98 g, 12 mmol) in glacial acetic acid (20 ml) was treated with zinc dust (5.2 g, 80 mmol), and the resulting suspension stirred vigorously at room temperature for 18 h. The mixture was then filtered through Celite and the residue washed with ethyl acetate (4 x 20 ml). The combined organic filtrate and washings were then extracted with water (200 ml portions) until the aqueous extracts did not effervesce on treatment with solid sodium hydrogen carbonate. The organic phase was dried and evaporated to give crude *2-acetyl-3-hydroxynaphthalene 17* as a yellow solid (1.79 g, 80%), m.p. 112 - 114 °C [lit.¹⁷ 110 - 112 °C (hexane)]; ν_{\max} 1655 cm⁻¹ (lit.¹⁷ 1651 cm⁻¹); δ (60 MHz) 2.7 (3 H, s, Me), 7.0 - 7.9 (5 H, m, ArH), 8.2 (1 H, s, 1-H), and 11.5 (1 H, s, OH).

4-Oxo-4H-1-naphtho[2,3-c]pyran-3-carboxaldehyde 4. - A solution of *2-acetyl-3-hydroxynaphthalene 17* (1.86 g, 10 mmol) in *N,N*-dimethylformamide (30 ml) at room temperature was treated with phosphorus oxychloride (4 ml), and the mixture stirred at 50 °C (bath) for 4 h. After cooling, the reaction mixture was poured on to ice water (100 ml) and stirred until a solid had formed. The solid was collected on a filter, washed with water (4 x 25 ml), and dried. The crude material was crystallised from ethyl acetate to give the *title compound 4* (1.33 g, 59%) as plates, m.p. 222 - 224 °C (lit.¹⁸ 220 - 222 °C); ν_{\max} 1690 and 1650 cm⁻¹; δ (80 MHz, d₆-acetone) 7.55 - 7.8 (2 H, m, ArH), 8.05 - 8.30 (3 H, m, ArH), 8.70 (1 H, s, 5-H), 8.80 (1 H, s, 2-H), and 10.25 (1 H, s, CHO).

Preparation of 3-(4-Chlorophenylsulphonyl)-4H-1-benzopyran-4-one 5 (Scheme 2).

2-(4-Chlorophenylthio)-2'-hydroxyacetophenone 20. - To a solution of 2-bromo-2'-hydroxyacetophenone **19**¹⁹ (3.2 g, 15 mmol) in dioxan (5 ml) was added dropwise at room temperature a solution of sodium 4-chlorothiophenoxide [from sodium (0.35 g, 15.2 mmol) and 4-chlorothiophenol (2.2 g, 15.2 mmol)] in absolute ethanol (10 ml). After stirring for a further hour the mixture was treated with water (30 ml) and the resulting precipitate collected by filtration. Washing with water, drying by suction, and crystallisation from dichloromethane - petroleum gave the *title compound 20* (3.5 g, 84%) as colourless needles, m.p. 83 - 84 °C (Found: C, 60.3; H, 4.1. C₁₄H₁₁ClO₂S requires C, 60.3; H, 4.0%); ν_{\max} 1625 and 1605 cm⁻¹; δ (60 MHz) 4.22 (2 H, s, CH₂), 6.7 - 7.9 (8 H, m, ArH), and 11.85 (1 H, s, exchanges with D₂O, OH); *m/z* 280 [M⁺ (³⁷Cl), 21%], 278 [M⁺ (³⁵Cl), 53], and 121 (100).

2-(4-Chlorophenylsulphonyl)-2'-hydroxyacetophenone 21. - To a stirred solution of the acetophenone **20** (1.40 g, 5 mmol) in dichloromethane (10 ml) was added dropwise at 0 °C a solution of 3-chloroperoxybenzoic acid (1.1 g, 6.4 mmol) in dichloromethane (20 ml). The mixture was kept at 0 °C for 5 h and water (30 ml) then added. The dichloromethane layer was separated and the aqueous layer extracted with dichloromethane (3 x 30 ml). The combined organic phase was washed successively with water (2 x 50 ml), saturated aqueous sodium hydrogen carbonate (50 ml), and brine. The solution was evaporated and the residue crystallised from ether - dichloromethane to give the pure *title compound 21* (1.36 g, 93%), m.p. 131 - 132 °C (Found: C, 57.3; H, 3.8. C₁₄H₁₁ClO₃S requires C, 57.05; H, 3.8%); ν_{\max} 1650, 1600, and 985 cm⁻¹; δ (60 MHz) 4.25 and 4.55 (each 1 H, d, *J* 14 Hz, CH₂), 6.7 - 7.8 (8 H, m, ArH), and 11.67 (1 H, s, exchanges with D₂O, OH); *m/z* 296 [M⁺ (³⁷Cl), 9%], 294 [M⁺ (³⁵Cl), 24], 248 (38), 246 (83), and 121 (100).

3-(4-Chlorophenylsulphinyl)-4H-1-benzopyran-4-one 5. - A mixture of the acetophenone **21** (1.17 g, 4 mmol), acetic-formic anhydride²⁰ (3.5 g, 40 mmol), and anhydrous sodium formate (2.8 g, 40 mmol) was heated to 70 - 75 °C for 3 h and allowed to stand at room temperature for a further hour. Water (20 ml) was then added and the mixture extracted with dichloromethane (3 x 20 ml). The combined extracts were washed with water and brine, dried, and evaporated to obtain the crude product, which was crystallised from ethanol to afford the pure *title compound 5* (0.72 g, 60%) as colourless crystals, m.p. 160 - 161 °C (Found: C, 59.2; H, 3.1. C₁₅H₉ClO₃S requires C, 59.1; H, 3.0%); ν_{\max} 1640, 1610, and 1050 cm⁻¹; δ (60 MHz) 7.25 - 8.25 (8 H, m, ArH) and 8.40 (1 H, s, 2-H); m/z 306 [M⁺ (³⁷Cl), 78%], 304 [M⁺ (³⁵Cl), 100], 257 (36), 255 (81), and 121 (93).

Cycloaddition of the Aldehyde 1 to 2,3-Dimethyl-1,3-butadiene. - To a stirred solution of chromone-3-carboxaldehyde **1** (200 mg, 1.15 mmol) in chloroform (10 ml) was added 2,3-dimethyl-1,3-butadiene (0.94 g, 11.5 mmol) followed by titanium(IV) chloride (2 drops). The mixture was heated under reflux for 96 hours and then concentrated to dryness *in vacuo*. The residue was chromatographed over silica gel [ether - petroleum, b.p. 60 - 80 °C, 1:9]. The major fraction contained *trans-2,3-dimethyl-1,4,4a,9a-tetrahydroxanthone 9* (61 mg, 23%), which formed colourless prisms, m.p. 148 - 150 °C (ether) (Found: C, 79.0, H, 7.2; C₁₅H₁₆O₂ requires C, 78.9, H, 7.1%); ν_{\max} (neat) 1680, 1600, and 1570 cm⁻¹; δ (90 MHz) 1.74 (6 H, s, 2 x Me), 2.5 - 2.7 (4 H, m, 1-H₂, 4-H₂), 2.85 (1 H, m, 9a-H), 4.40 (1 H, ddd, *J* 8, 8, 13 Hz, 4a-H), 6.9 - 7.1 (2 H, m, 5-H, 7-H), 7.48 (1 H, ddd, *J* 2, 8, 8.5 Hz, 6-H), and 7.90 (1 H, dd, *J* 2, 8.5 Hz, 8-H); m/z 229 (M + 1, 14%), 228 (M⁺, 58), 213 (61), 198 (38), 195 (87), and 121 (100). Another fraction contained *cis-2,3-dimethyl-1,4,4a,9a-tetrahydroxanthone 10* (47 mg, 18%), which was an oil, b.p. 115 - 120 °C (0.5 mmHg); ν_{\max} (neat) 1680, 1600, and 1575 cm⁻¹; δ (60 MHz) 1.7 (6 H, s, 2 x Me), 2.05 - 3.1 (5 H, m, 1-H₂, 4-H₂, 9a-H), 4.8 (1 H, narrow m, 4a-H), 6.9 - 7.2 (2 H, m, 5-H, 7-H), 7.55 (1 H, ddd, *J* 2, 8, 8 Hz, 6-H), and 8.0 (1 H, dd, *J* 2, 8 Hz, 8-H).

Cycloaddition of the Ester 3 to 2,3-Dimethyl-1,3-butadiene. - To methyl 4-oxo-4H-1-benzopyran-3-carboxylate **3** (218 mg, 1.07 mmol) in dichloromethane (20 ml) was added 2,3-dimethyl-1,3-butadiene (820 mg, 10 mmol) followed by one drop of titanium(IV) chloride. The brown solution was monitored by t.l.c. [ethyl acetate - petroleum (1:5)], and when the ester was no longer detectable, the mixture was poured into water (25 ml) and the aqueous layer extracted with dichloromethane (2 x 20 ml). The combined organic phase was dried and evaporated. Flash chromatography of the residue [ethyl acetate - petroleum (1:6)] gave *methyl cis-1,4-dihydro-9-oxo-3-[[trimethylsilyl]oxy]xanthene-9a-carboxylate 11* (257 mg, 84%) as a colourless oil (Found: C, 71.1, H, 6.5; C₁₇H₁₈O₄ requires C, 71.3, H, 6.3%); ν_{\max} (neat) 1740 and 1685 cm⁻¹; δ (60 MHz) 1.7 (6 H, br s, 2 x Me), 2.2 - 2.6 (4 H, m, 1-H₂, 4-H₂), 3.75 (3 H, s, CO₂Me), 5.1 (1 H, t, *J* 4 Hz, 4a-H), 6.9 - 7.2 (2 H, m, 5-H, 7-H), 7.4 - 7.7 (1 H, m, 6-H), and 7.95 (1 H, dd, *J* 2, 8 Hz, 8-H).

General Procedure for the Cycloadditions of 1, 3, and 4 to the Diene 12. - To a magnetically stirred solution of **1** or **3** (or a suspension of **4**) (1.0 mmol) in dichloromethane (5 ml) was added the diene **12** (5 mmol). To monitor the reactions, aliquots were removed periodically, concentrated *in vacuo* (room temperature, \leq 0.5 mmHg), and the residual oil examined by ¹H n.m.r. spectroscopy. Under these conditions the aldehydes **1** and **4** were consumed within 6 - 8 h, whereas the ester **3** was still detectable [δ (90 MHz) 8.78 (s, 2 H)] after 72 - 96 h. In each case the signals characteristic of methoxy and/or aldehyde groups indicated that two cycloadducts had been formed in ratios \leq 1.5:1.

The mixture of cycloadducts **13** had δ (60 MHz) 0.2 (total 9 H, 2 x s, SiMe₃), 2.5 - 2.8 (2 H, br m, 4-H₂), 3.1 and 3.3 (total 3H, both s, OMe), 4.6 and 4.75 (total 1 H, both d, *J ca.* 6 Hz, 1-H), 5.0 - 5.7 (2 H, m, 2-H, 4a-H), 6.9 - 8.1 (4 H, m, ArH), and 9.9 and 10.0 (total 1 H, both s, CHO); ν_{\max} (neat) 1730 and 1675 cm⁻¹.

The mixture of cycloadducts **14** had δ (90 MHz) 0.3 (2 x s, SiMe₃), 2.5 - 2.6 (br m, 4-H₂), 3.1 - 4.3 (m, OMe), 4.65 and 4.8 (both d, *J ca.* 6 Hz, 1-H), 5.1 - 5.8 (m, 2-H, 4a-H), and 6.9 - 8.0 (m, ArH); ν_{\max} (neat) 1735 and 1675 cm⁻¹.

The mixture of cycloadducts **18** had δ (60 MHz) 0.2 (total 9 H, 2 x s, SiMe₃), 2.5 - 2.8 (2 H, br m, 4-H₂), 3.2 and 3.4 (total 3H, both s, OMe), 4.75 and 4.9 (total 1 H, both d, *J ca.* 5.5 Hz, 1-H), 5.1 - 5.8 (2 H, m, 2-H, 4a-H), 7.3 - 8.1 (5 H, m, ArH), 8.7 (1 H, s, 11-H), and 10.05 and 10.15 (total 1 H, both s, CHO); ν_{\max} (neat) 1725 and 1675 cm⁻¹.

Cycloaddition of 5 to the Diene 12. - A solution of 3-(4-chlorophenylsulphinyl)-4H-1-benzopyran-4-one **5** (250 mg, 0.92 mmol) and the diene **12** (90%; 1.0 ml, 4.6 mmol) in 1,2-dimethoxyethane (10 ml) under nitrogen was heated under reflux for 18 h, cooled, and stirred at room temperature with M hydrochloric acid (2 ml) for 2 h. The mixture was then treated carefully with a solution of sodium hydrogen carbonate (0.3 g) in water (3 ml), and evaporated. The residue was

extracted with dichloromethane (3 x 15 ml), washed with water (2 x 20 ml), and dried. The residue on evaporation of the solution was triturated with dichloromethane - petroleum (1:2), and the solid collected and dried *in vacuo*, giving essentially pure 3-hydroxy-9H-xanthen-9-one **23** (87.5 mg, 50%), m.p. 253 - 254 °C (ethanol - water) (lit.²¹ 246 °C); *m/z* 213 (M + 1, 14%), 212 (M⁺, 100), 184 (19), 177 (19), 128 (10), 92 (17), and 63 (11); R_f 0.14 [ethyl acetate - dichloromethane (1:19)].

Desilylation of Cycloadducts 13 using Tetrabutylammonium fluoride. - Tetra-*n*-butylammonium fluoride in tetrahydrofuran (1.0 M; 1.1 ml, 1.1 mmol) was added dropwise to a stirred solution of the adducts **13** (347 mg, 1.0 mmol) in dichloromethane (10 ml), and the resulting reddish solution stirred at room temperature for 24 h. The solvents were removed *in vacuo* and the residue purified by flash chromatography [ethyl acetate - petroleum (1:3)] to give a green oil which solidified on standing. Recrystallisation gave pure 2,4'-dihydroxybenzophenone **24** (94 mg, 44%) as colourless plates, m.p. 148 - 150 °C (acetic acid - water) (lit.²² 150 - 151 °C); ν_{\max} 3400 - 3200, 1625, 1595, 1250, 1165, 940, 855, 835, 765, 725, and 690 cm⁻¹; δ (300 MHz) 5.66 (1 H, s, 4'-OH), 6.86 (1 H, ddd, *J* 1.0, 7.0, and 8.0 Hz, 5-H), 6.91 (2 H, br d, *J* 8.6 Hz, 3'-H₂), 7.04 (1 H, dd, *J* 1.0, 8.4 Hz, 3-H), 7.47 (1 H, ddd, *J* 1.7, 7.0, and 8.4 Hz, 4-H), 7.60 (1 H, dd, *J* 1.7, 8.0 Hz, 6-H), 7.64 (2 H, dm, *J* 8.6 Hz, 2'-H₂), and 11.94 (1 H, s, 2-OH); R_f 0.52 [ethyl acetate - dichloromethane (1:9)].

Desilylation of Cycloadducts 13 using HF in Acetonitrile. - A solution of the cycloadducts **13** (from 1.0 mmol of **1**) in acetonitrile (5 ml) was stirred at room temperature with a solution of aqueous hydrofluoric acid (40% HF; 0.1 ml) in acetonitrile (5 ml) for 1 h. The mixture was then concentrated *in vacuo* to give an orange oil, which was chromatographed on silica gel [dichloromethane - ethyl acetate (24:1)]. A yellow fraction containing several components was triturated with ether - petroleum (2:1) to give a colourless solid, identified as (1 α ,4 $\alpha\beta$,9 $\alpha\beta$)-1,2,4 α ,9 α -tetrahydro-1-methoxy-3,9-dioxo-4H-xanthen-9 α -carboxaldehyde **25** (122 mg, 45%), m.p. 120 - 125 °C (dec.) [ethyl acetate - petroleum (b.p. 60 - 80 °C)] (Found: C, 65.8; H, 5.2. C₁₅H₁₄O₅ requires C, 65.7; H, 5.15%); ν_{\max} 1730, 1674, and 1607 cm⁻¹; δ (300 MHz) 2.44 (1 H, dd, *J* 3.7, 15.0 Hz, 2-H_{ax}), 2.83 (1 H, overlapping ddd, *J* ~ 1, 2.9, and 15.0 Hz, 2-H_{eq}), 2.91 (2 H, br d, *J* 3.5 Hz, 4-H₂), 3.09 (3 H, s, OMe), 4.51 (1 H, overlapping ddd, *J* 1.9, 2.9, and 3.7 Hz, 1-H), 5.17 (1 H, overlapping ddd, *J* 1.9, 3.5, and 3.5 Hz, 4 α -H), 6.95 (1 H, d, *J* 8.6 Hz, 5-H), 7.02 (1 H, overlapping dd, *J* 7.0, 7.8 Hz, 7-H), 7.47 (1 H, overlapping ddd, *J* 1.7, 7.0, and 8.6 Hz, 6-H), 7.85 (1 H, dd, *J* 1.7, 7.8 Hz, 8-H), and 9.98 (1 H, s, CHO). On decoupling at 4.51 ppm, the signal at 5.17 ppm becomes a triplet (*J* 3.5 Hz), and the signals at 2.44 and 2.83 ppm lose their couplings of 3.7 and 2.9 Hz respectively; on decoupling at 5.17 ppm, the signal at 4.51 ppm becomes an overlapping doublet (*J* 2.9 and 3.7 Hz), and the signal at 2.91 ppm becomes a broadened singlet.

Conversion of 25 to 2,4'-dihydroxybenzophenone 24. - The adduct **25** (42 mg, 0.153 mmol) and anhydrous sodium acetate (20 mg) in ethanol (2 ml) were heated to boiling for 15 min and allowed to cool. The mixture was concentrated *in vacuo* and the residue suspended in ethyl acetate. Filtration of the solution through a short column of silica gel, washing with more ethyl acetate, and concentration of the eluate gave 2,4'-dihydroxybenzophenone **24** (33 mg, 100%), identical (t.l.c., n.m.r., m.p.) with the material obtained as described above.

Desilylation of Cycloadducts 14. - To a solution of the ester adducts **14** (from 1.0 mmol of **3**) in tetrahydrofuran (5 ml) was added aqueous hydrochloric acid (0.1 M; 0.1 ml) and the mixture stirred at room temperature for 20 h. Water (10 ml) was then added, and the solution extracted with dichloromethane (3 x 10 ml). The combined organic phase was dried, evaporated, and the residue chromatographed [ethyl acetate - petroleum (1:5)]. A fluorescent fraction was found to contain methyl (1 β ,4 $\alpha\beta$,9 $\alpha\beta$)-1,2,4 α ,9 α -tetrahydro-1-methoxy-3,9-dioxo-4H-xanthen-9 α -carboxylate **26** (68 mg, 22%), m.p. 137 - 142 °C (dec.) (ethyl acetate - petroleum) (Found: C, 63.15; H, 5.3. C₁₆H₁₆O₆ requires C, 63.15; H, 5.3%); ν_{\max} 1730 and 1690 cm⁻¹; δ (80 MHz) 2.65 - 2.95 (4 H, m, 2-H₂, 4-H₂), 3.28 (3 H, s, OMe), 3.72 (3 H, s, CO₂Me), 4.59 (1 H, t, *J* 3.5 Hz, 1-H), 5.47 (1 H, dd, *J* 7.5, 10 Hz, 4 α -H), 6.95 (2 H, t, *J* 8 Hz, 5-H, 7-H), 7.47 (1 H, dt, *J* 2, 8 Hz, 6-H), and 7.79 (1 H, dd, *J* 2, 8 Hz, 8-H).

Desilylation of Cycloadducts 18. - A solution of the cycloadducts **18** (from 1.0 mmol of **4**) in acetonitrile (5 ml) was stirred at room temperature with a solution of aqueous hydrofluoric acid (40% HF; 0.1 ml) in acetonitrile (5 ml) for 1 h. The mixture was then concentrated *in vacuo* to give an orange-red oil, which was purified by chromatography [dichloromethane

- ethyl acetate (24:1)]. A fluorescent product (R_f 0.4) was eluted in two fractions; the first crystallised on evaporation to give essentially pure (1 β ,4 α ,12 α)-1,2,4a,12a-tetrahydro-1-methoxy-4H-benzo[b]xanthene-3,12-dione **27** (93 mg); the next fraction also contained a slightly more polar yellow component which was removed by trituration with ether, giving a further sample (40 mg) of **27** (total 133 mg, 45%). The analytical sample, pale yellow needles, had m.p. 168 - 170 °C (dec.) (ethyl acetate) (Found: C, 73.0; H, 5.5. $C_{18}H_{16}O_4$ requires C, 73.0; H, 5.4%); ν_{max} 1725, 1690, 1630, 1605, 1210, 1190, 1135, 1095, 1000, 945, 890, and 760 cm^{-1} ; δ (300 MHz) 2.42 (1 H, dd, J 2.8, 15.4 Hz, 2- H_{ax}), 2.81 (1 H, dd, J 11.2, 14.5 Hz, 4- H_{ax}), 2.91 (1 H, overlapping ddd, J 2.1, 2.5, and 15.4 Hz, 2- H_{eq}), 2.94 (1 H, dd, J 2.1, 12.5 Hz, 12a-H), 3.10 (1 H, ddd, J 2.1, 5.6, and 14.5 Hz, 4- H_{eq}), 3.31 (3 H, s, OMe), 4.59 (1 H, overlapping ddd, J 2.1, 2.5, and 2.8 Hz, 1-H), 4.89 (1 H, ddd, J 5.6, 11.2, and 12.5 Hz, 4a-H), 7.31 (1 H, s, 6-H), 7.34 (1 H, ddd, J ca.1, 6.8, and 8.1 Hz, 9-H), 7.49 (1 H, ddd, J ca.1, 6.8, and 8.2 Hz, 8-H), 7.68 (1 H, br. d, J 8.2 Hz, 7-H), 7.86 (1 H, br. d, J 8.1 Hz, 10-H), and 8.52 (1 H, s, 11-H). Line broadening on the signals for 7-H and 10-H derives from combined small couplings to 6-H and 8-H, and 9-H and 11-H respectively; m/z (methane CI) 298 ($M + 2$, 19.9%), 297 ($M + 1$, 100.0), 295 ($M - 1$, 14.7), 293 (5.1), 265 (16.7), 197 (6.1), and 171 (11.7); (EI) 296 (M^+ , 100%), 223 (27), 210 (47), 170 (64), 142 (34), 114 (24), and 85 (42).

ACKNOWLEDGEMENTS

We thank the S.E.R.C. and Glaxo Group Research (Greenford) for a C.A.S.E. award (to P.J.C.), the British Council for a Postgraduate Studentship (to S.T.S.), and Drs. Stan Roberts and Mike Gregson (Glaxo) for their contributions to this work.

REFERENCES

- 1 For examples, see T.W. Wallace, *Tetrahedron Lett.*, 1984, 25, 4299; P.D. Clarke, A.O. Fitton, H. Suschitzky, T.W. Wallace, H.A. Dowlatshahi, and J.L. Suschitzky, *ibid.*, 1986, 27, 91; P.J. Cremins and T.W. Wallace, *J. Chem. Soc., Chem. Commun.*, 1984, 1698; P.J. Cremins and T.W. Wallace, *ibid.*, 1986, 1602.
- 2 S.T. Saengchantara and T.W. Wallace, *J. Chem. Soc., Chem. Commun.*, 1986, 1592.
- 3 S.T. Saengchantara and T.W. Wallace, *J. Chem. Soc., Perkin Trans. 1*, 1986, 789, and references cited therein.
- 4 For examples, see G.P. Ellis, 'Chromenes, Chromanones and Chromones,' Wiley, New York, 1977, ch. 7; C.K. Ghosh, *J. Heterocycl. Chem.*, 1983, 20, 1437; B. Chantegrel, A.I. Nadi, and S. Gelin, *J. Org. Chem.*, 1984, 49, 4419.
- 5 3-Substituted chromones appear to function as dienophiles in some dimerisation reactions; see C.K. Ghosh, A. Bhattacharyya, and C. Bandyopadhyay, *J. Chem. Soc., Chem. Commun.*, 1984, 1319.
- 6 For examples, see J.W. Lown, S.M. Sondhi, S.B. Mandal, and J. Murphy, *J. Org. Chem.*, 1982, 47, 4304; J.L. Charlton, V.A. Sayeed, and G.N. Lypka, *Can. J. Chem.*, 1982, 60, 1996; C.-M. Wong, W. Haque, H.-Y. Lam, K. Marat, E. Bock, and A.-Q. Mi, *ibid.*, 1983, 61, 1788; J.W. Lown and S.M. Sondhi, *J. Org. Chem.*, 1984, 49, 2844; and references cited therein.
- 7 For a synthesis and leading references, see N. Katagiri, J. Nakano, and T. Kato, *J. Chem. Soc., Perkin Trans. 1*, 1981, 2710.
- 8 R.F. Newton, D.P. Reynolds, M.A.W. Finch, D.R. Kelly, and S.M. Roberts, *Tetrahedron Lett.*, 1979, 3981.
- 9 R.C. Gupta, D.A. Jackson, R.J. Stoodley, and D.J. Williams, *J. Chem. Soc., Perkin Trans. 1*, 1985, 525.
- 10 D.D. Perrin, W.L.F. Armarego, and D.R. Perrin, 'Purification of Laboratory Chemicals,' 2nd Edition, Pergamon, Oxford, 1980.
- 11 W.C. Still, M. Khan, and A. Mitra, *J. Org. Chem.*, 1978, 43, 2923.
- 12 H. Harnisch, *Justus Liebigs Ann. Chem.*, 1972, 765, 8.
- 13 Y. Machida, S. Nomoto, S. Negi, H. Ikuta, and I. Saito, *Synth. Commun.*, 1980, 10, 889.
- 14 L.F. Fieser and M. Fieser, 'Reagents for Organic Synthesis,' Volume 1, Wiley, New York, 1967, p. 668.
- 15 S. Klutchko, M.P. Cohen, J. Shavel, and M. von Strandtmann, *J. Het. Chem.*, 1974, 11, 183.
- 16 J.B. Cohen and H.W. Dudley, *J. Chem. Soc.*, 1910, 97, 1732.
- 17 I.M. Hunsberger, *J. Am. Chem. Soc.*, 1950, 72, 5626.
- 18 M. von Strandtmann and S. Klutchko, U.S. Patent 3 886 183 (1975).
- 19 M.L. Malik and S.K. Grover, *Indian J. Chem., Sect. B*, 1976, 14, 513.
- 20 L.I. Krimen, *Org. Synth.*, 1970, 50, 1.
- 21 H. Atkinson and I.M. Heilbron, *J. Chem. Soc.*, 1926, 2688.
- 22 A. Baeyer, *Justus Liebigs Ann. Chem.*, 1907, 354, 177.